## Heterocyclic Synthesis with 3-Cyano-2(1*H*)pyridinethione: Synthesis of 3-Oxo-2,3-dihydroisothiazolo[5,4-b]pyridine and Related Compounds [1]

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**Summary.** 3-Carbethoxy-4,6-diphenyl-2-pyrridine sulfonamide (5), can be cyclized to 3-oxo-2,3-dihydro-4,6-diphenylisothiazolo[5,4-b]pyridine-1,1-dioxide (2). Oxidation of pyridinethione 6 with  $Cl_2/H_2O$  gave the sulfonyl chloride derivative 7, which can be ammonolyzed to 3-amino-4,6-diphenylisothiazolo[5,4-b]pyridine-1,1-dioxide (8), and 3-cyano-4,6-diphenylpyridine-2-sulfonamide (9). Hydrolysis of 6 gave 3-carboxamido-2(1*H*)pyridinethione (12) which can be oxidized with iodine to 3oxo-4,6-diphenyl-2,3-dihydroisothiazolo[5,4-b]pyridine (13). 3-Methyl-4,6-diphenylisothiazolo[5,4b]pyridine-1,1-dioxide (17) was also prepared from 6.

Keywords. 2-Amino-4,6-diphenylethyl nicotinate; 3-Cyano-4,6-diphenyl-2-mercaptopyridine.

### Heterocyclensynthese mit 3-Cyano-2(1*H*)pyridinthion: Synthese von 3-Oxo-2,3-dihydroisothiazolo-[5,4-b]pyridin und verwandten Verbindungen

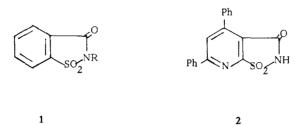
**Zusammenfassung.** 3-Carbethoxy-4,6-diphenyl-2-pyridinsulfonamid (5) kann zu 3-Oxo-2,3-dihydro-4,6-diphenylisothiazolo[5,4-b]pyridin-1,1-dioxid (2) cyclisiert werden. Die Oxidation des Pyridin-thions 6 mit  $Cl_2/H_2O$  ergab das Sulfonylchlorid-Derivat 7, das mit Ammoniak zu 3-Amino-4,6-diphenylisothiazolo[5,4-b]pyridin-1,1-dioxid (8) und 3-Cyano-4,6-diphenylpyridin-2-sulfonamid (9) umgesetzt werden kann. Die Hydrolyse von 6 ergab 3-Carboxamido-2(1*H*)pyridinthion (12), das mit Jod zu 3-Oxo-4,6-diphenyl-2,3-dihydroisothiazolo[5,4-b]pyridin (13) oxidiert wurde. 3-Methyl-4,6-diphenyl-isothiazolo[5,4-b]pyridin-1,1-dioxid (17) wurde ebenfalls aus 6 hergestellt.

### Introduction

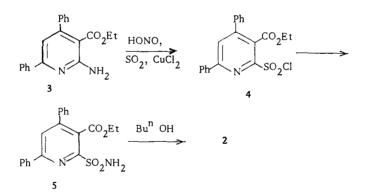
The considerable biological activity of 1,2-benzisothiazole-3(2H)-one 1,1-dioxide (1) as fungicide, antiviral and antibacterial agent [2] has prompted considerable interest in synthesis and chemistry of these compounds. As a part of a programme directed for developing new compounds for utility as antischitosomal agent we became interested in synthesizing derivatives of 1 in which the benzene ring is replaced by a nitrogen heteroaromatic ring.

### **Results and Discussion**

The precursors 2-amino-4,6-diphenylethyl nicotinate (3) and 3-cyano-4,6-diphenyl-2-mercapto pyridine (6) were synthesized following the literature procedure [3]. The 2-NH<sub>2</sub> group of 3 can be replaced, thus on treatment with nitrous acid in presence of cold saturated solution of SO<sub>2</sub> in acetic acid containing copper(II)



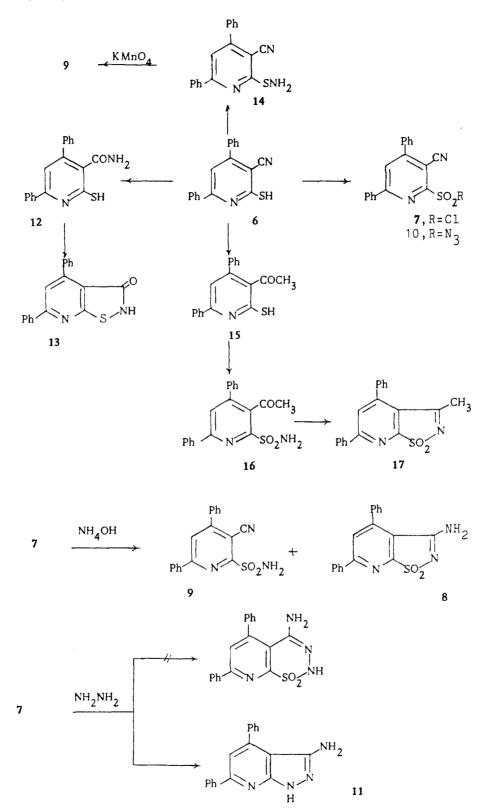
chloride dihydrate [4], it gives the sulfonyl chloride derivative 4, which on direct ammonolysis with ammonium hydroxide afforded the sulfonamide derivative 5 in 50% yield. The structure of 5 was established based on the IR spectrum which showed absorption bands at  $3400-3300 \text{ cm}^{-1}$  (v NH<sub>2</sub>), at  $1700 \text{ cm}^{-1}$  (v C=O ester), and at 1320,  $1150 \text{ cm}^{-1}$  typical for a SO<sub>2</sub> group. The sulfonamide 5 was cyclized in boiling butanol to give 3-oxo-2,3-dihydro-4,6-diphenylisothiazolo[5,4b]pyridine-1,1-dioxide (2). The <sup>1</sup>H NMR of 2 showed the presence of an 11 H multiplet centered at  $\delta 8.00-7.66$  due to the aromatic protons, and a broad weak singlet at  $\delta 11.00$  due to the NH proton.



Since this synthesis was successful, it was of interest to synthesize substituted isothiazolo[5,4-b]pyridine-1,1-dioxide derivatives. It has been found that when **6** was treated with chlorine gas in presence of 10% acetic acid at 0°C the corresponding sulforyl chloride **7** was obtained, which on direct ammonolysis with NH<sub>4</sub>OH [5] afforded 3-amino-4,6-diphenylisothiazolo[5,4-b]pyridine-1,1-dioxide (**8**) and 3-cy-ano-4,6-diphenylpyridine-2-sulfonamide (**9**).

On treatment with sodium azide in aqueous acetone the sulfonyl chloride 7 was readily converted into the corresponding sulfonyl azide 10 [6].

Similarly, treatment of 7 with anhydrous hydrazine gave 3-amino-4,6-diphenylpyrazolo[3,4-b]pyridine (11) instead of pyridothiadiazine-1,1-dioxide [7]. The formation of 11 is the consequence of nucleophilic attack on the C-2 by the chlorine atom followed by hydrazine attack. The IR revealed the presence of an  $NH_2$  group



 $(3400-3100 \text{ cm}^{-1})$ , no bands due to either the cyano group or the sulfone group were detectable.

Schaper [8] has reported the synthesis of 3-oxo-2,3-dihydroisothiazolo[5,4b]pyridines by oxidation reactions of 3-carboxamido-2(1*H*)pyridinethione. We have prepared 3-carboxamido-4,6-diphenyl-2(1*H*)pyridinethione (12) using acidic hydrolysis of 6 [9]. Compound 12 was treated with iodine in benzene at reflux temperature, 3-oxo-2,3-dihydroisothiazolo[5,4-b]pyridine (13) was obtained in 60% yield. The structure of 13 was confirmed by ir by the presence of bands at 3400, 1650, and 1100 cm<sup>-1</sup> due to (NH), (C=O), and (-S-) groups.

The amination of 6 was investigated, thus treatment of 6 with sodium hypochlorite followed by ammonium hydroxide gave 2-sulfenamide derivative 14, which can be oxidize by potassium permanganate giving the corresponding sulfonamide 9.

On the other hand, the reaction of 6 with methylmagnesium iodide afforded 3-acetyl derivative 15 [10]. When 15 was treated with chlorine gas in acetic acid at 0°C it gave the corresponding sulfonyl chloride, which on direct ammonolysis with ammonium hydroxide afforded the sulfonamide derivative 16 in 50% yield.

The sulfonamide **16** was cyclized in boiling acetic anhydride giving 3-methyl-4,6-diphenylisothiazolo[5,4-b]pyridine-1,1-dioxide (**17**). The structure of **17** was confirmed by ir, the presence of a bands at 1 370, 1 200 cm<sup>-1</sup> being typical of an SO<sub>2</sub> group. No C=O band was observed in the ir spectrum of compound **17**. The <sup>1</sup>H NMR spectrum exhibited lines at  $\delta 8.20$ -7.4 (m, 11 H, aromatic protons), and 2.5 (s, 3 H, CH<sub>3</sub>).

Extension of the synthetic scope of these reactions are under study.

### **Experimental Part**

Melting points are uncorrected. <sup>1</sup>H NMR spectra were determined in  $DMSO-d_6$  on a Varian 60 MH<sub>2</sub> spectrometer with TMS as internal standard and chemical shifts are expressed as (ppm). IR spectra were obtained in a Pye Unicam Sp-1100 spectrophotometer (KBr). Analytical data were performed by the Microanalytical centre, Cairo University.

### 3-Carbethoxy-4,6-diphenyl-2-pyridinesulfonamide (5)

2-Amino-4,6-diphenylethyl nicotinate (3) (3.18 g, 10 mmol) in acetic acid (20 ml) and concentrated HCl (10 ml) was treated with sodium nitrite (1.16 g) at 0°C and the diazonium salt solution poured all at once into a cold saturated solution of  $SO_2$  in acetic acid (15 ml) containing copper(II) chloride dihydrate (1 g). The solid product obtained was filtered off and washed with water to give the sulfonyl chloride 4; this was added to ammonium hydroxide (150 ml). The reaction mixture was stirred at 5–10°C for 2 h, the resulting solution was concentrated to half of its volume and neutralized with hydrochloric acid to precipitate 5 (see Table 1).

### 3-Oxo-2,3-dihydro-4,6-diphenylisothiazolo[5,4-b]pyridine-1,1-dioxide (2)

Compound 5 (3.82 g, 10 mmol) was refluxed in butyl alcohol (30 ml) for 3 h, excess solvent was removed and the residue set aside for 24 h. Then solid **2** was collected by filtration.

Compound	M.p. °C Solvent	IR cm <sup>-1</sup>	<sup>1</sup> H NMR (ppm)	Yield %	Formula <i>m</i> /e	Analysis % Calc./Found	%
	of cryst.*					С	Н
0	125 E	3 300 (NH), 3 200 (OH), 1 700 (C=O), 1 350, 1 100 (SO <sub>2</sub> )	11 (br, 1 H, NH), 8.00–7.66 (m, 11 H, <i>Ar</i> )	57	C <sub>18</sub> H <sub>12</sub> N <sub>2</sub> O <sub>3</sub> S 336.36	64.27 64.20	3.59 3.40
Ŋ	$\frac{220}{E}$	3 400–3 300 (NH <sub>2</sub> ), 1 700 (C = O, ester), 1 320, 1 150 (SO <sub>2</sub> )	8.30–7.40 (m, 11 H, <i>Ar</i> ), 6.3 (br, 2 H, NH <sub>2</sub> ), 4.4 (9, 2 H, CH <sub>2</sub> ), 2.8 (s, 3 H, CH <sub>3</sub> )	37	C <sub>20</sub> H <sub>18</sub> N <sub>2</sub> O <sub>4</sub> S 382.42	62.81 62.80	4.74 4.90
×	176 $E$	3 300–3 250 (NH <sub>2</sub> ), 1 320, 1 150 (SO <sub>2</sub> )	I	60	$C_{18}H_{13}N_3O_2S$ 335.36	64.46 64.29	3.90 4.10
6	220 DMF	3 100 (NH), 2 200 (CN), 1 320, 1 150 (SO <sub>2</sub> )	8.20–7.40 (m, 11 H, <i>Ar</i> ), 6.2 (br, 2 H, NH <sub>2</sub> )	31	$C_{18}H_{13}N_3O_2S$ 335.36	64.46 64.60	3.90 4.20
10	170 $E$	2 220 (CN), 2 175 (N <sub>3</sub> ), 1 350, 1 150 (SO <sub>2</sub> )	1	28	C <sub>18</sub> H <sub>11</sub> N <sub>5</sub> O <sub>2</sub> S 361.37	59.82 60.20	3.06 3.20
11	$\frac{210}{E}$	3400-3100 (NH <sub>2</sub> ), 1610 (C = N)	8.32–7.27 (m, 12 H, <i>Ar</i> ), 4.5 (br, 2 H, NH <sub>2</sub> )	60	$C_{18}H_{14}N_4$ 286.32	75.50 75.60	4.92 5.00
12	140 $E$	$3400 (NH_2), 1660 (C=O),$ 1100 (-S-)	8.50–7.40 (m, 11 H, <i>Ar</i> ), 6.20 (br, 2 H, NH <sub>2</sub> ), 3.73 (m, 1 H, SH)	82	C <sub>18</sub> H <sub>14</sub> N <sub>2</sub> OS 306.37	70.56 70.30	4.60 4.30
13	250 M	3 200 (NH), 1 650 (C = O), 1 100 $(-S-)$	9.8 (br, 1H, NH), 8.20–7.6 (m, 11H, Ar)	60	C <sub>18</sub> H <sub>12</sub> N <sub>2</sub> OS 304.36	71.02 71.10	3.97 3.80
14	220 $E$	3 200 (NH <sub>2</sub> ), 2 210 (CN), 1 600 (C=N)	8.3-7.5 (m, 11 H, Ar), 6.10 (br, 2 H, NH <sub>2</sub> )	55	$C_{18}H_{13}N_3S$ 303.36	71.26 71.10	4.31 4.20
16	135 E	3 400–3 300 (NH <sub>2</sub> ), 1 650 (C=O), 1 320, 1 150 (SO <sub>2</sub> )	8.10–7.6 (m, 11 H, <i>Ar</i> ), 6.00 (br, 2 H, NH <sub>2</sub> ), 2.7 (s, 3 H, CH <sub>3</sub> )	51	C <sub>19</sub> H <sub>16</sub> N <sub>2</sub> O <sub>2</sub> S 352.40	64.75 64.20	4.57 4.30
17	190 B/P	1 370, 1 200 (SO <sub>2</sub> )	8.20-7.4 (m, 11 H, Ar), 2.5 (s, 3 H, CH <sub>3</sub> )	50	C <sub>19</sub> H <sub>14</sub> N <sub>2</sub> O <sub>2</sub> S 334.38	68.24 68.70	4.21 4.10

\* B Benzene, E Ethanol, M Methanol, P Petrolether (40–60°), DMF dimethylformamide

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# Table 1. Analytical and spectral data

# 3-Amino-4,6-diphenylisothiazolo[5,4-b]pyridine-1,1-dioxide (8) and 3-cyano-4,6-diphenylpyridine-2-sulfonamide (9)

In a solution of 3-cyano-4,6-diphenyl-2(1*H*)pyridinethione (6) (2.88 g, 10 mmol) in acetic acid (100 ml, 10%), chlorine gas was bubbled at 0°C. A white precipitate started forming. After 1 h the precipitate was filtered, washed with water several times, and dried to give 2-sulfonylchloride derivative 7; this was added to conc.  $NH_4OH$  (100 ml). The reaction mixture was heated on a steam bath for 4 h, the volume was reduced to the half and neutralized with 6*N* HCl to give a solid product. The solid product was washed with water, dried, and then treated with acetic acid. It was divided into two parts: an acetic acid soluble part and an acetic acid insoluble part. The acetic acid solution was concentrated to obtain 8. The acetic acid insoluble part afforded 9.

### 3-Cyano-4,6-diphenylpyridine-2-sulfonylazide (10)

Sodium azide (1.3 g, 20 mmol) in water (10 ml) was added to a stirred solution of 7 (3.55 g, 10 mmol) in acetone (50 ml) and the solution was stirred at room temperature for further 10 h. It was evaporated in vacuo to one third of its volume and water (200 ml) was added. The precipitate of 10 was filtered, washed with water and dried.

### 3-Amino-4,6-diphenylpyrazolo[3,4-b]pyridine (11)

To a solution of 7 (3.55 g, 10 mmol) in absolute ethanol (30 ml) was added anhydrous hydrazine (10 mmol). The solution was heated under reflux for 6 h. It was evaporated to one-third of its volume. Then 11 precipitated from the solution.

### 3-Carboxamido-4,6-diphenyl-2(1H)pyridinethione (12)

Compound 6 (2.88 g, 10 mmol) was dissolved in conc.  $H_2SO_4$  (30 ml) and heated on a steam bath for 3 h. The solution was cooled to room temperature and poured over crushed ice (200 g). The precipitate of 12 was filtered and dried.

### 3-Oxo-2,3-dihydro-4,6-diphenylisothiazolo[5,4-b]pyridine (13)

To a solution of 12 (1 g, 3.2 mmol) in benzene (20 ml) was added a solution of iodine (1.27 g, 5 mmol) in benzene (10 ml) and the solution was heated under reflux for 5 h. Then excess iodine was destroyed with a saturated solution of potassium hydroxide and precipitated 13 was filtered.

### 3-Cyano-4,6-diphenylpyridine-2-sulfenamide 14

6 (2.88 g, 10 mmol) was added to a solution of sodium hypochlorite (20 ml, 5 mmol). The reaction mixture was stirred at 30°C for 3 h. Ammonium hydroxide (50 ml) was added to the reaction mixture, the mixture was stirred for an additional hour at room temperature, the precipitate of 14 was filtered and washed with water several times.

### Oxidation of 14

To a solution of 14 (3.03 g, 10 mmol) in acetone (20 ml) was added a solution of potassium permanganate (3.28 g, 20 mmol) in water (60 ml) portion-wise and the solution was stirred at room temperature for 12 h. The excess permanganate was destroyed with sulfur dioxide and the solution filtered to remove the precipitated manganese dioxide. The aqueous solution was made strongly alkaline with sodium hydroxide pellets (2 g) and then extracted with chloroform ( $3 \times 20$  ml). Evaporation of the dried (Na<sub>2</sub>SO<sub>4</sub>) chloroform extract gave 9 (1.89 g, 56%). The compound had a melting point and spectral properties identical to those of a sample obtained in the previous experiment.

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### 3-Acetyl-4,6-diphenylpyridine-2-sulfonamide (16)

To a cold solution of 3-acetyl-4,6-diphenyl-2(1*H*)pyridinethione (15) (3.05 g, 10 mmol) in dilute acetic acid (100 ml, 10%) chlorine gas was bubbled at 0°C. A white precipitate started forming. After 1 h the precipitate was filtered, washed with water several times and dried; this 2-sulfonyl chloride derivative was added to conc. ammonium hydroxide (100 ml). The reaction mixture was heated on a steam bath for 3 h and then neutralized with hydrochloric acid to give 16.

### 3-Methyl-4,6-diphenylisothiazolo[5,4-b]pyridine-1,1-dioxide (17)

A solution of 16 (3.52 g, 10 mmol) in acetic anhydride (20 ml) was heated under reflux for 3 h. The reaction mixture was cooled and a saturated solution of sodium bicarbonate was added. Solid 17 was separated by filtration.

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